

Industrial Processing of Cooked Red Cabbage Pieces (*Brassica Oleracea L.*)

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Abstract: Red cabbage may contribute to human health due to the contents of anthocyanins and the compounds released from these colorants that also may be very important for the quality characteristics of the processed red cabbage. The quality characteristics of the red cabbage produced from three industrial companies were equal because the factories are using a few cultivars and very ingredient composition and almost the same glass jar sizes. Red cabbage contains several odour compounds with very attractive odour characteristics that may be described further and applied for processing of several different foods using a variety of cuttings in combination with variation in brine composition on raw and processed cabbages. The contents of anthocyanins may improve the quality characteristics of cut and canned red cabbage foods significantly. Using a variety of red cabbage cultivars and variation in fertilizer composition, sowing times according to different sums of degree days may improve the quality characteristics of the red cabbage. The most value of canned red cabbage foods may be obtained using the variously health improving compounds with significantly and efficient properties as health promoting compounds.

Keywords: red cabbage, cooking, colour, harvest time and health promotion.

1. INTRODUCTION

The Danish food industries are processing cooked red cabbage and may appreciate that red cabbages are known as a very healthy vegetable because it conveys a variety of health benefits, depending on absorption and metabolic mechanisms that deliver anthocyanins and their bioactive metabolites to responsive red cabbage tissues [1]. Research in brassica species have shown that the anthocyanins from red cabbage have an inhibitory potential against digestive enzymes linked to obesity by inhibition of enzymes involved in the carbohydrate and lipid digestion [2]. These anthocyanins are well known for their toxic effects in both man and animals at high doses. In contrast at sub toxic doses, their hydrolytic and metabolic products acts as chemo protective agents against chemically induced carcinogens by blocking of the initiation of tumor developments in a variety of tissues, including liver, colon, mammary gland and pancreas [3]. Besides are those carcinogenic compounds known as glucosinolates that may result in blocking of Phase I and Phase II biotransformation enzymes and suppression of tumours by apoptosis [3,4,5]. Recent research have shown that the active anthocyanins in red cabbage have an inhibitory potential against digestive enzymes linked to obesity by inhibition against enzymes involved in carbohydrate and lipid digestion including α -amylase, α -glucosidase, and lipase [6]. Subtoxic doses of hydrolytic and metabolic compounds produced in the red cabbage tissue are supposed to act as bio protective agents towards chemically-induced carcinogens because they are blocking the tumor initiation in a variety of tissues, including liver, colon, mammary gland and pancreas as examples. These compounds exhibit their effects by inducing phase I and phase II enzymes, inhibiting the enzyme activation, modifying the steroid hormone contents and protection against significantly oxidative damage [3]. In that connection may the anthocyanins in red cabbage be considered as a very rich source of antioxidants as presented previously [7]. Cooking and blanching of red cabbage may result in severe reductions in the contents of aliphatic and indole glucosinolates by about certitive percent, respectively. The contents of ascorbic acid, total phenols, anthocyanins may be reduced significantly by causing 19, 15, 38, and 28 percent losses during processing of these compounds in the red cabbage tissues. Besides may these compounds cause significantly reductions in glucosinolates and antioxidant related parameters in red cabbage [8]. And therefore are the aim of this research to study the changes in the contents of minerals, sugars, odour and anthocyanins during processing of red cabbage pieces.

2. MATERIALS AND METHODS

The raw materials for processing of red cabbage in this research were obtained using cabbage heads from the cultivar 'Aurore' delivered from the three participating companies experienced in growing and processing of red cabbage. Company one used a blancher with shower head for supply of hot blanching water and companies two and three used steam blanchers with Archimedean screws. Company two did not use pumping at this stage of processing. Processing of red cabbage on the three industrial processing lines a1, a2 and a3 were carried out according to normal practise at the factories using red cabbage heads from the cultivar 'Aurore'. Processing of red cabbage in this research included cutting of the cabbage leaves into pieces < 5 mm in length, cooking in an open nor closed pot, cooling of the cabbage pieces in cold water to 30°C, packing in glass jars, size 530 cm³ jars, pasteurized at 80°C for 20 min, cooled in cold water and stored at 12°C in a cooling cabinet until further analysis and sensory evaluation. The contents of dry matter were determined by drying of 200 g macerated samples at 80°C for twenty hours in a heating cabinet and drained weight was determined after actual storage time using a standard sieve (DIN 4188) with 2.5 mm sieve openings. The contents of potassium were measured using a ten times diluted standard solution with 0.55 g Ca(OH)₂ and 54.75 g CaCl₂, 6 g H₂O l⁻¹. The obtained standard for calculation of the potassium contents in g potassium were: g L⁻¹ = - 0.0456mv -1.106 r = 0.99). Soluble solids were measured w/w using a refractometer and titratable acid were determined by titration of macerated samples to pH 8.1 using 0.1 N NaOH in water. The anthocyanins were extracted using 0.01 w/w % HCl and the contents of anthocyanins were measured in duplicates at 530 nm using a Beckmann spectrophotometer. Measurement of surface colour included determination lightness L, redness a and yellowness b using a Hunter colorimeter. Firmness were measured using a Kramer-shear cell packed with 175 g cabbage materials and stem velocity 20 mm min⁻¹. The contents of volatiles and carbohydrates were extracted and measured as described previously [8]. And the contents of volatiles were collected in 500 mL distillate at 40°C obtained by disintegration of 2000 g cabbage with 2000 g distilled water in a blender followed by distillation of 500 ml (12.5%) at 40 °C into from the mixture of this mixture and distillation of volatiles were contents of volatiles were contents of odour compounds were collected by mixing of two kg cooked water with two kg distilled water using a Waring blender and transferred to a distillation column in order to collect 500 ml red cabbage distillate containing 95 per cent of the volatiles applicable for sensory evaluation by sniff analysis in connection with sensory evaluation as described previously [9]. The size of the glass jars was 570 cm³ and they were packed with 280 g cooked cabbage and 250 g brine kept at 70°C. Pasteurization was carried out in water baths kept at 80 °C for 30 min and thereafter were the filled jars cooled in tap water at 60°C for 5 min and in cold water for 20 min before further analyses and evaluations were carried out. Drained weight was measured using a sieve (DIN 4188, 2.5 cm). The samples of canned red cabbage were stored at 15°C for six months and the average drained (dw) weight decreased significantly linearly with cooking time $dw = 313 - 0.123min$ r = 0.98. The sensory quality characteristics of processed red cabbage were evaluated by an intensively trained panel of six women and three men age 22-25 year and trained for several years in sensory evaluation of several fruits and vegetables consumed in fresh or processed stage. Potassium were measured using an ion specific electrode (Corning) in samples diluted with a solution of CaCl₂ and Ca(OH)₂ as described by the supplier of the equipment. The obtained equation for calculation of the potassium contents in g potassium g l⁻¹ = - 0.0456mv -1.106 r = 0.99. The anthocyanins were extracted using 0.01 w/w % HCl and the contents of anthocyanins were measured in duplicates (*Brassica oleracea L.*). Samples of the processed red cabbage were evaluated by an intensively trained panel of five women and two men age 22-25 trained in four years sensory evaluation of several fruits and vegetables consumed as fresh or processed stage.

3. RESULTS AND DISCUSSION

The average content of dry matter after cutting increased significantly from factory a1 to factory a3 after cutting and were at maximum for factory a2 after blanching (Table 1). The contents of dry matter increased significantly from factory a1 to factory a3 after cutting, were non significantly different after pumping and significantly different between the factories after blanching. Soluble solids were at maximum for a3 after cutting and none significantly different in the other cases. Potassium was only significantly lowest after pumping and blanching for company a3. The contents of anthocyanin were significantly lowest for company a3 after cutting, highest after pumping for a3 and. highest for a2 after blanching. Previous studies of the effects of blanching showed that the losses were highest for

Chinese cabbage 40% after 15 min of blanching, 27% for white cabbage, while the losses for Chinese white cabbage and red cabbage were 19.9 and 4.0%, respectively [9]. Composition of the raw cut cabbages processed by three companies (a₁, a₂, a₃) processing red cabbage using the cultivar ‘Auto-ro’ with no specific pumping at factor

Table1. Composition of the raw cabbage from processing at the three industrial companies.

Treatments	Dry matter g 100 ⁻¹			Soluble solids g 100g ⁻¹			Potassium mg 100g ⁻¹			Anthocyanin mg 100 g ⁻¹		
	a ₁	a ₂	a ₃	a ₁	a ₂	a ₃	a ₁	a ₂	a ₃	a ₁	a ₂	a ₃
After	a ₁	a ₂	a ₃	a ₁	a ₂	a ₃	a ₁	a ₂	a ₃	a ₁	a ₂	a ₃
Cutting	9.1c	9.8b	10.1a	5.4b	5.2b	6.8a	2.4a	2.4a	2.1a	17.0a	17.3a	16.5b
Pumping	7.7a		7.5a	4.3a		4.6a	2.0a		1.7b	10.4a		14.1b
Blanching	8.2c	9.1a	8.7 b	5.1a	4.9a	5.1a	1.8a	1.8a	1.6b	9.8c	14.7b	13.5a

This research included studies in the possibilities for application of blanched red cabbage juice as an ingredient by processing of red cabbage with the aim to improve the sensory properties of cooked red cabbage (Table 2). The contents of anthocyanin were not significantly different after cutting and significantly different between the factories after blanching. The contents of sugars were significantly different after cutting with maximum for a₃ and non significance after pumping and blanching for a₁ and a₃. Total dry matter, anthocyanin, sugar and potassium in the blanching juices were significantly different between the three factories at several points.

Table2. Contents of anthocyanins, sugar and dry matter on the three industrial processing lines.

Treatments	Anthocyanin mg 100 g ⁻¹			Sugar g 100 g ⁻¹			Blanching juice composition.			
	a ₁	a ₂	a ₃	a ₁	a ₂	a ₃	Compounds	a ₁	a ₂	a ₃
After cut- ting	170a	173a	165a	5.4b	5.2b	6.8a	Total dry matter, g 100 ⁻¹	3.7b	4.7a	3.9b
After pumping	133b	-	141a	4.3a		4.6a	Anthocyanin, mg 100 g ⁻¹	73.4c	101.1a	103.4a
After blanching	98c	147a	135b	5.1a	4.9a	5.1a	Sugar, mg 100 g ⁻¹	25.1b	35.7a	19.4c
After can- ning	114a	-	114a	22.0b		25.7a	Potassium, mg 100 ⁻¹	2.0b	2.1b	6.8a

After packing with sugar was the most important increases of minor importance, whereas the losses increased significantly from 6.8 to 25.7 g 100 g⁻¹. The contents of potassium were significantly lowest for company a₃ and similar for company a₁ and a₂. The average losses of potassium were about 25 w/w % and the total losses in soluble solids varied from 6.1 to 24.1 g 100 g⁻¹. On the basis of the data in table 2 it was concluded that processing resulted in significant losses of dry matter, potassium, anthocyanin and sugars. After pumping on the three processing lines were the contents of dry matter reduced significantly. The contents of dry matter increased significantly both during blanching and especially by packing that resulted in 7.1 to 14.3 g 100 g⁻¹ losses in dry matter. The changes in sugar varied from 4.2 to 25.7 percent. Immediately after cutting of the cabbage heads were the contents of anthocyanin in the cut and packed cabbage heads non-significantly different between the three companies (Table 2). The contents of anthocyanin in the cabbage heads were significantly different after pumping and blanching, but not after packing. The contents of sugar were significantly different after cutting, but not in other cases. The differences in total dry matter, anthocyanin, sugar and potassium were significantly different. Some varieties accumulated more than 30 per cent acylated pigments, and the proportions of mono-acylated pigments may decrease with processing time [9] The data in table 3 shows that the contents of volatiles measured by gas chromatography changed significantly during processing with more than and the non-concentrated blanching juice that contained 14.8 mg kg⁻¹ anthocyanin. The contents of reducing sugars varied significantly from 6.1 to 24.1 g 100 g⁻¹. Removal of the odour compounds and variation in the soluble solids, acids and anthocyanins may increase the possibilities for application of blanching juice for improvement of the sensory quality characteristics of red cabbage. The odour of the volatile compounds that were eluted by gas chromatography was evaluated by members of the sensory panel. Seven compounds through the table with retention time 20, 22, 42, 54, 114, 116, 136 and 165 min were odourless. The odour of compounds eluted after 25, 40, 45, 68, 118 and 144 min had an odour as freshly cut or raw grasses that may contribute signifi-

cantly positively to the fresh odour of red cabbage heads. Odour as butter, vanilla, cinnamon 30, 34, 58, 90, 119, 130 and sweet cabbage may contribute positively to the odour of cooked red cabbage. On the contrary may compounds eluted after 84 and 85 min be characterized as strong off odours and 71, 75 and 130 min as raw cabbage and 144 was characterized as hay. The data in table 3 showed that red cabbage contained eight compounds without odour, five compounds with odour as fresh cut grass, four with odour as butter, one as vanilla, two as cinnamon, three as raw cabbage, one sweet cabbage, one sharp and one irony as found by the eight members of the intensively trained sensory panel with eight participants. The data in table 3 shows that red cabbage contained eight compounds without odour, five compounds with odour as fresh cut grass, four with odour as butter, one as vanilla, two as cinnamon, three as raw cabbage, one sweet cabbage, one sharp and one irony as found by the eight members of the intensively trained sensory panel with eight participants. Sensory analysis of the eluates by gas chromatography after 20, 22, 42 min, showed that the surface colours obtained from company a₂ encompassed thertien samples with surface colour +a₂red 18-23 and 25 samples from +a₃ red 10-22. Besides occurred three synthetic samples in position +a₃red 10-12. The samples from +a₃ had a more intensive colour in comparison to samples from company +a₂, that may be due to application red currant juices in both processing lines.

Table3. Odour compounds in cooked red cabbage with retention time and detector counts.

Odour	Min	Raw	Blanched	Cooked	Min	Odour	Raw	Blanched	Cooked
none	20	1001b	2514a	2522a	75	raw cabbage	1436a	491c	691b
“	22	no	no	624a	84	sharp	744b	1339a	no
fresh cut grass	25	26c	46b	102a	85	iron/tin	no	no	439a
butter	30	no	no	909a	90	mustard	23	no	no
vanilla	34	no	835b	17618a	114	not known	2937a	752b	no
fresh cut grass	40	448a	27b	no	116	none	no	no	339
none	42	563a	no	no	118	fresh grass	432a	172b	no
fresh cut grass	45	814a	no	68b	119	sweet cabbage	no	no	1519a
none	54	272c	783b	25283a	130	raw cabbage	2303a	516b	585b
cinnamon	58	1503b	1843a	710c	136	none	471c	560b	9264a
fresh cut grass	68	1464a	738b	326c	144	hay	144b	18.6c	3312a
red cabbage	71	no	no	929a	165	none	1025a	803b	313c

The other compounds with decreasing level during processing included compounds with retention time 30, 42, 75, 90, 114, 118, 130 and 165 min and several compounds with increasing level during processing included 20, 22, 30, 34, 71. The odour of the compounds eluted after 40, 42, 45, 68, 75, 114, 118, 130 and 165 min was mainly characterized as freshly cut green leaves and the concentration decreased with elution time. Compounds with increasing level during processing included the compounds with retention time 20, 22, 25, 30, 34, 54, 71, 81, 85, 116, 119, 136 and 144 min. Table 4 shows the data from odour disruption of volatile compounds detected by the sensory panel by sniffing to the eluates at the end of the separation column as described previously by separation of volatiles in green beans [8]. The contents of compounds without odour were eluted after 20, 22, 114 and 116 min. The aim of this study was also to investigate the effects of different home cooking techniques, boiling, steaming, and stir-frying in red cabbage, on the levels of anthocyanins and phenolic compounds determined by high-performance liquid chromatography coupled with photodiode array and mass spectrometry detectors and on the antioxidant activity evaluated by and cellular antioxidant activity assays. The steaming technique resulted in significant increases in phenolic content in kale whereas in red cabbage it was significantly reduced. In the kale, steaming resulted in significant increases in antioxidant activity levels in all of the evaluation methods. In the red cabbage, boiling resulted in a significant increase in antioxidant activity using the abts assay. According to the CAA assay, the stir-fried sample displayed the highest levels of antioxidant activity [10].

Table4. Number of colour spots of red cabbage from company a₂ and a₃.

Factory	+a ₂ red							Synthetic juice	Currant juice
a ₂	< 18	18-19	19-20	20-21	21-22	22-23	Sum		yes
Red spot	1	3	2	4	2	1	13		
	+a ₃ red								
a ₃	10-12	12-14	14-16	16-18	18-20	20-22		10-12	yes
Red spots	2	6	3	5	4	1	25	3	

Thermal degradation of individual glucosinolates within the plant matrix was studied. Red cabbage samples were heated at different temperatures for various times. To rule out the influence of enzymatic breakdown and to focus entirely on the thermal degradation of glucosinolates, myrosinase was inactivated prior to the thermal treatments. All identified glucosinolates degradation when heated at temperatures above 100 °C. The indole glucosinolates 4-hydroxy-glucobrassicin and 4-methoxy-glucobrassicin showed the highest degree of showed degradation, even at temperatures below 100 °C. Kinetic parameters have been estimated for the degradation that could be described by first-order kinetics. At temperatures below 110 °C indole glucosinolates have a significant higher degradation rate constant as compared to aliphatic glucosinolates. The breakdown of 4-hydroxy-gluco-brassicin seems to consist of two parallel reaction pathways. Based on the proposed degradation kinetics and the estimated parameters, the degree of thermal degradation of all individual glucosinolates at standardized heating conditions (blanching, cooking and canning) was simulated. Glucosinolates are expected to be not very susceptible to thermal degradation during blanching conditions. Cooking will cause more thermal degradation to indole glucosinolates (38%) as compared to aliphatic glucosinolates (8%). Canning, the most severe heat treatment, will result in significant thermal degradation (73%) of the total amount of glucosinolates [11].

Table5. Effects of processing at the two companies on quality characteristics.

Company	Measures	Firmnes, kg	Anthocyanin mg 100 g ⁻¹	Lightness	Blueness	Redness
a ₃	Aerage	109.0a	35.5a	17.4a	15.0a	-0.5d
a ₃	Std. dev.	29.7c	8.49b	1.2d	2.8d	0.9a
b ₂	Average	9.7c	35.9a	15.9b	12.7b	-0.9
b ₂	Std. dev.	35.5b	12.9c	4.4c	3.7c	1.0c

A previous study aimed to evaluate the anthocyanin pigment contents and profiles from seven red cabbage cultivars at two maturity stages eight weeks apart regarding their color characteristics and behavior under acidic and neutral ph. The contents of anthocyanin concentrations ranged from 1111 to 1780 mg cyanidin-3-glucoside in 100 g dry matter and the contents did not increase further with growing time new [12]. Cultivar and maturation affected the pigment profile. Some varieties accumulated >30% of diacylated pigments, and proportions of monoacylated pigments decreased with time. Extracts from selected varieties at first harvesting time produced colors similar to the extracts from the second harvest with higher proportion of diacylation. Cultivar selection and maturation affected color and stability of red cabbage extracts at different pH values [13].

Table6. Average of quality characteristics for samples from factory a₂ and a₃ (n = 24)

	Firmness, kg	Anthocyanine mg 100 g ⁻¹	Lightness L	Greenness a	Yellowness b	Firmness kg	Anthocyanine mg 100 g ⁻¹	Lightness L	Greenness a	Yellowness b
	Factory a ₂					Factory a ₃				
Avg.	119.6	33.0	17.6	16.2	0.1	202.4	36.1	18.6	20.2	-1.4
Std.	24.8	6.8	1.0	2.4	0.5	47.7	4.5	0.8	1.4	0.3
23	864	92	0	59	4	20	948	1	29	4
1	243	4	0	11	0	2	446	0	2	0

The data in table 6 shows the average of quality characteristics of canned red cabbage at factory a₂ and a₃ by increasing and constant firmness, respectively. The average and standard deviations are typically for the processed samples from the two companies. It may be remarked that there occur significantly differences in yellowness of slices from the two companies. Humans are unable to synthesize L-ascorbic acid (l-aa), ascorbate, vitamin C, and are thus entirely dependent upon dietary sources to

meet the needs. In both plant and animal metabolism, the biological functions of l-ascorbic acid are centered on the antioxidant properties of this molecule. Considerable evidence has been accruing in the last two decades of the importance of l-aa in protecting not only the plant from oxidative stress, but also mammals from various chronic diseases that have their origins in oxidative stress. Evidence suggests that the plasma levels of l-aa in large sections of the population are sub-optimal for the health protective effects of this vitamin [13]. During cooking and blanching are the glucosinolates degraded according to first order kinetics with rate constants that varied between four to twenty fold between the vegetables [14]. Previous research showed that blanching of red cabbage resulted in significantly reductions in phenols, monomeric anthocyanins, ferric reducing ability power, oxygen-radical absorbance and ascorbic acid and soluble sugar contents [15]. Boiling resulted in less extensive reductions, while steaming caused reduced phenols and soluble sugars. However, significant reductions were found for monomeric anthocyanins and ascorbic acid. In general were losses accounted for in the processing waters, however the losses in total monomeric anthocyanins were not fully recovered, indicating degradation. Total glucosinolates were severely affected by processing, with degradation of 64, 38 and 19 percent in blanched, boiled and steamed red cabbage while aliphatic and indole glucosinolates were similarly affected and the lost parts were partially recovered in the processing water [15]. The data in table 7 shows the significantly differences in the contents of anthocyanin between the three cultivars and throughout the picking period for 'Auroro'. The three columns in the right part of table 6 shows that the red cabbage firmness decreased significantly and linearly with the blanching time. The contents of anthocyanin of the red cabbage cultivars measured from medium September to medium December showed significantly different contents of anthocyanin from 51 to 100 mg 100 g⁻¹, whereas the contents of dry matter varied significantly between months (Table 7). The average drained weight (Dw) decreased significantly linearly with cooking time $Dw = 313 - 0.123min$. $r = 0.98$ and the average firmness decreased exponentially with cooking time. Until quite recently, little focus has been given to improving the l-aa-metabolism content of plant foods, either in terms of the amounts present in commercial crop varieties, or in minimizing losses prior to ingestion. Further, while l-aa biosynthesis in animals was elucidated in the 1960s, it is only very recently that distinct biosynthetic routes for plants have been proposed. The characterization of this new pathway will undoubtedly provide the necessary focus and impetus to enable fundamental questions on plant l-aa metabolism to be resolved.

Table7. Processing of red cabbage pieces using cabbage heads from three cultivars.

Cultivar	Picking	Anthocyanin	Dry matter	Anthocyanin	ln c	b	r
	Date	mg 100 g ⁻¹	g 100 g ⁻¹	mg/100 g dm			
Marne Sep.	13 th Sep	51a	2.57b	1984e	7.44a	-0.95a	0.95a
Ruby Perfec.	"	51f	3.23a	1578f	6.39c	-0.84b	0.99a
Auroro	27 th Sep	126a	3.22a	3913a	6.15b	-0.70c	0.97a
"	11 th Oct	99d	2.95a	3356c	6.70b	-0.69c	0.94a
"	20 th Nov	121b	3.06a	3954a	6.10d	-0.65c	0.96a
"	28 th Nov	111c	3.05a	3639b	6.10e	-0.59d	0.96a
"	10 th Dec	87e	3.19a	2727d			
"	13 th Dec	110c	3.06a	3595b			
Avg.		118	3.04	3093	6.48	- 0.74	0.96

This review focuses on the role of l-aa in metabolism and the latest studies regarding its biosynthesis, tissue compartmentalization, turnover and catabolism. These relationships are considered in relation to the potential to improve the l-aa test content of crops. Finally the factors that determine the bioavailability of l-aa test and how it may be improved are considered, as well as the most important future research needs [16]. During the food production chain from sowing, picking, storage and processing may the composition of glucosinolates change significantly. Blanching and cooking of vegetables led to considerable ($P < 0.05$) losses of total glucosinolates from 2.7 to 30.0% and from 35.3 to 72.4%, respectively. No systematic changes in total glucosinolates were found in the vegetables that

were blanched, frozen and stored for 48 h. The highest concentration of cancer-protective compounds, such as aliphatic and indole glucosinolates were found in Brussels sprouts (sinigrin and glucobrassicin) and in broccoli (glucoraphanin) [12]. The average of quality characteristics of samples of red cabbage processed at factory a₂ and a₃ showed significantly differences between and within and every property that occur because of the variation in maturity and development in the cabbage heads. Conventional cooking did not affect aliphatic glucosinolates significantly, while the indole glucosinolates decreased to a higher extent [13]. A large variation of the glucosinolate contents has been found previously by [14]. These results show that all cooking treatments, except steaming, caused significant losses of chlorophyll and vitamin C and significant decreases of total soluble proteins and soluble sugars. Total aliphatic and indole glucosinolates were significantly modified by all cooking treatments but not by steaming. In general, the steaming led to the lowest loss of total glucosinolates, while stir-frying and stir-frying/boiling presented the highest losses. Stir-frying and stir-frying/boiling may be the two most popular methods for most homemade dishes in that may cause great losses of chlorophyll, soluble protein, soluble sugar, vitamin C, and glucosinolates, but the steaming method appears to be the best in retention of the nutrients in cooking broccoli. The various methods used by processing of red cabbage may result in severe significantly reductions of bioactive compounds including carotenoids, anthocyanins and phenolic compounds [15].

Table8. *Drained weight and firmness of cultivars in dependence of cooking time.*

Cultivar	No	Cooking time, min											
		Drained weight, g 100 g ⁻¹						Firmness, kg					
		20	30	40	50	60	Avg.	20	30	40	50	60	Avg.
Marne	1	307b	308b	305b	310b	305a	307a	103.7a	67.7b	42.5c	42.0d	37.7e	58.7b
Ruby Perf.	1	311a	314a	308a	306c	305a	309a	48.0a	34.3b	27.c0	21.6d	19.3e	30.1f
Auroro	1	309b	304b	310a	314a	308a	309a	60.7a	44.3b	33.7c	32.3d	28.3e	39.9
“	2	311a	308b	311a	306c	307a	308a	108.0a	70.7b	70.7b	51.7c	48.3d	69.9a
“	1	312a	312a	308a	309c	306a	309a	78.3a	56.7b	49.0c	43.3d	38.7e	53.2d
“	2	312a	308b	309a	307c	308a	309a	85.3a	57.7b	49.7c	44.7d	42.3e	55.9c
“	1	308b	311a	308a	301d	304b	306a	67.8a	46.7b	40.0c	37.3d	32.0e	44.7e
“	2	315a	312a	312a	305e	302b	309a	72.3a	57.0b	52.3c	43.7d	41.3e	53.3d

Cooking techniques may improve the levels of bioactive compounds and antioxidant activity in red cabbage. Blanching and frozen storage of cauliflower resulted in significantly decreases in aliphatic and indole glucosinolates by 31 and 37 per cent and the contents of l-ascorbic acid decreased 19 per cent, total phenols 15, whereas anthocyanins were reduced 28 per cent. The effect of thermal treatment of glucosinolates and antioxidant-related parameters in red cabbage was found by [9]. The cooking water from the normal processing of red cabbage were rich in anthocyanins, soluble solids and organic acids and may generally be used for cooking of red cabbage instead of using pure water as shown in table 10. The starting levels were 24.5 mg 100 g⁻¹ soluble solids, 0.75 g 100 g⁻¹ acidity, 0.75 and anthocyanin 10.9 mg 100 g⁻¹. The results from nine more cooking are shown in table 10. Anthocyanin in cooking water increased from zero to 38.8 mg 100 g⁻¹ and shows that the soluble solids increased from 24.5 to 26.5 mg 100 g⁻¹, acidity increased from 0.75 to 0.79 g 100 g⁻¹, anthocyanin from 10.9 to 18.4 mg 100 g⁻¹, lightness and greenness decreased whereas yellowness, ΔE and firmness increased significantly and pH was constant. After nine cooking were soluble solids and anthocyanin in the red cabbage increased from 3.4 to 7.8 g 100 g⁻¹ and the contents of anthocyanine increased from 24.0 to 40.5 mg 100 g⁻¹. The average contents of soluble solids were 24.9, acidity 0.76 and anthocyanin were 26.8 mg 100 g⁻¹. The equations in table 9 showed that greenness and yellowness increased with the levels of acetic acid in the produced red cabbage and that firmness decreased with processing time. The contents of anthocyanin and yellowness decreased and increased with processing time, respectively.

Table9. Changes in quality characteristics including acetic acid, soluble solids and processing time.

Firmness = 100.8 - 16.5 acetic acid	0.81	$\ln(\text{kg}) = 7.76 - 0.72 \ln(\text{min} + 1) - 0.012\text{min}$	0.97
$a = 5.1 + 6.9 \text{ acetic acid}$	0.96	Anthocyanin = 42.3 - 0.17min	0.93
$b = -1.33 + 2.2 \text{ acetic acid}$	0.99	Yellowness = - 0.40 + 0.0189 min	0.80
$\text{kg} = 139.7 - 1.27 \text{ soluble solids}$	0.95		

Increasing cooking time from 10 to 60 min did not affect the contents sugar, pH or acetic acid in the red cabbage, whereas anthocyanin and firmness decreased significantly with the cooking time. It was found that blanching and cooking of the vegetables led to significantly losses of total glucosinolates from 2.7 to 30.0 per cent and from 35.3 to 72.4 per cent, respectively [16]. The contents of cabbage and lightness were not affected by increasing cooking time, while greenness decreased and yellowness increased significantly. The effects of thermal treatment on glucosinolates and antioxidant-related parameters in red cabbage are described previously [9]. As pointed out will blanching and cooking of red cabbage result in severe reductions of aliphatic and indole glucosinolates by 31 and 37 percent and the contents of ascorbic acid, total phenols, anthocyanins by blanching [17]. The contents of glucosinolates may be classified as aliphatic, aromatic, omega-methylthioalkyl and heterocyclic compounds [3]. Studies in the effects of blanching, cooking and canning on the contents of eight glucosinolates showed that the rate of degradation was lowest after blanching, medium after cooking and maximum after canning. [17]. The glucosinolates have been described as responsible for the characteristic flavour and odour compounds in red cabbage foods [18]. That are because these compounds are released from the red cabbage heads due to myrosinase activity and enzymatic hydrolysis of the [19]. The products of these intermediate compounds in dependence of pH and substrate availability of ferrous ions [15]. Carisa; Mercadante, Adriana. Optimization of cooking techniques may improve the levels of bioactive compounds and antioxidant activity in red cabbage. The effects of domestic processing affect the occurrence of glucosinolates randomly [20], [21], whereas the industrial processes has been less studied [22, 23, 24] and it is generally assumed that the responsible dietary constituents includes vitamins, minerals, dietary fibre, polyphenols and polyphenolics. The next experiment with cooking of red cabbage were carried out by transfer in two series using from 0 to 900 and from 900 to 0 g processing water. The contents of anthocyanin increased significantly from zero to 38.8 mg 100 g⁻¹ using the extra supply of the compounds during processing. The contents of soluble solids, acidity were constant 38.8 mg 100 g⁻¹ and 9.0 g 100 g⁻¹ soluble solids, respectively. Soluble solids, acidity, lightness, pH were constant while anthocyanin in the processed cabbage increased significantly. Blueness, redness, ΔE and the contents The contents of

Table10. Data from processing of red cabbage pieces by reusing of the cooking water.

Anthocyanin mg 100 g ⁻¹	Soluble solids g 100 g ⁻¹	Acidity g 100 g ⁻¹	Lightness	Anthocyanin mg 100 g ⁻¹	Blueness	Redness	ΔE	Firmness kg	pH	Anthocyanin mg 100 g ⁻¹
0.0j	24.5a	0.8a	27.1a	10.9d	16.7d	1.4a	0f	75g	3.8a	24.0f
4.3i	25.3a	0.8a	26.1a	15.6c	18.2a	1.4a	2.4e	70f	3.7a	34.3e
8.6h	26.0a	0.8a	24.4a	15.6b	17.2b	1.4a	3.2d	84d	3.5a	34.3d
12.9g	26.4a	0.8a	23.5a	17.7a	17.7c	2.3a	4.5c	78e	3.8a	38.9bc
17.2f	26.6a	0.8a	22.7a	18.7a	16.6d	1.5ab	4.7c	79e	3.8a	41.1a
21.5e	26.6a	0.8a	21.0a	17.2b	15.4e	2.7b	6.9b	80d	3.8a	36.8b
25.8d	26.9a	0.8a	21.6a	17.7b	14.4f	2.9b	6.7b	91c	3.9a	34.5b
30.1c	26.4a	0.8a	22.9a	18.1a	13.9g	3.4c	6.2b	96b	3.8a	39.8a
34.4b	26.6a	0.8a	21.7a	18.0a	14.3h	3.6c	7.0a	99a	3.9a	35.2b
38.8a	26.5a	0.8a	21.3a	18.4a	13.5i	2.9b	7.3a	99a	3.9a	40.5a

The changes in quality characteristics using increasing blanching water resulted in a higher content of soluble solids and reduced firmness whereas the contents of all other compounds were non-significantly different (Table 10). A very detailed description of the effects of blanching, boiling and steaming on glucosinolates, total phenols, total anthocyanins, L-ascorbic acid, antioxidant potential, ferric reducing ability power ferric reducing ability power, oxygen radical absorbance capacity, individual native glucosinolates and losses by blanching, boiling and steaming was found by [9]. Red cabbage leaves are rich in glucosinolates that may be classified as aliphatic, aromatic, ω-methylthioalkyl and heterocyclic compounds [3], that are of special importance because they have anti-carcinogenic properties that may result in blocking of phase I and phase II biotransformation enzymes and suppress tumours by apoptosis [4]. Myrosinase or thioglucoside glucohydrolase (E.C. 3.2.3.1) is the trivial name for the enzyme or group of enzymes that catalyses the hydrolyse of glucosinolates [18] Fenwich and Heaney 1983. The mechanical damage produced by slicing of the red cabbage leaves resulted in mixing of the enzyme molecules with glucosinolates and a row of intermediates that rearranged and resulted in production of compounds in dependence of the composition of the red cabbage leaves such as pH, substrate or availability of minerals [19]. The products from hydrolysis of glucosinolates include isothiocyanates, nitriles, thiocyanates indoles and oxazolidinethiones [4, 24]

Table11. Composition of red cabbage cooked by increasing levels of soluble solids in the cooking water.

Soluble solids g 100 g ⁻¹	pH	Acidity g 100 g ⁻¹	Antho- cyanin mg 100 g ⁻¹	Firmness kg	L	Blueness	Redness	ΔE
14.3a	3.9a	0.8a	18.2a	108a	24.3a	18.1a	-0.6a	-0.6a
17.0b	3.9a	0.8a	19.8a	107a	23.8a	17.5a	0.5a	0.5a
20.0c	3.9a	0.8a	16.7a	104b	23.7a	18.7a	-0.4a	-0.4a
22.0d	3.9a	0.8a	18.2a	100c	21.8a	17.0a	-1.0a	-1.0a
25.3e	3.8a	0.8a	22.3a	97d	22.9a	19.5a	0.0a	0.0a
28.4f	3.9a	0.8a	21.3a	94d	21.5a	19.6a	0.6a	0.6a
30.4g	3.9a	0.8a	20.8a	91d	22.9a	18.9a	0.2a	0.2a
34.2h	3.8a	0.8a	20.4a	92d	22.5a	19.5a	0.3a	0.3a
37.4j	3.9a	0.8a	19.7a	97d	22.2a	17.4a	0.7a	0.7a

The contents of isothiocyanates and indoles have been implicated to have anticarcinogenic properties [4]) and the data in table 7 shows that increasing heating time by processing of cooked red cabbage resulted in significantly decreases in redness only and the data in table 10 shows that increases in acetic acid in the brine not affected the sugar content or lightness, whereas the increasing levels of acetic acid resulted in higher levels in acidity, redness and yellowness, whereas firmness and pH decreased significantly. Increasing cooking time from 10 to 60 min did not affect the contents sugar, pH or acetic acid in the red cabbage, whereas anthocyanin and firmness decreased significantly with the cooking time. The contents of cabbage and lightness were not affected by increasing cooking time, while blueness decreased and yellowness increased significantly [6]. Increasing cooking time from 10 to 60 min did not affect the contents of soluble solids, pH or acetic acid in the red cabbage, whereas anthocyanin and firmness decreased significantly with the cooking time . The contents of cabbage and lightness were not affected by increasing cooking time, while blueness decreased and redness increased significantly [2].

Table12. Effects of increasing contents of sugar.

Sugar content	pH	Acetic acid g 100 g ⁻¹	Anthocyanin mg 100 g ⁻¹	Firmness kg	Cabbage g 100 g ⁻¹	Lightness	Blueness	Redness
9.6	4.0a	0.7a	37.4a	131.0a	62.6a	15.6a	10.5a	-3.1f
13.9	4.0a	0.7a	37.4a	118.4b	63.4a	15.5a	9.9b	-2.7e
17.8	4.0a	0.6a	36.6a	114.4c	65.2a	14.6a	9.2c	-2.1d
22.4	4.0a	0.6a	35.5a	112.2d	65.0a	15.9a	9.7d	-1.8c
26.0	4.0a	0.6a	35.8a	110.0e	64.6a	15.5a	9.6e	-2.2b
30.1	4.0a	0.6a	34.6a	100.0f	66.2a	15.7a	9.6e	-1.6a

Cooking will cause more thermal degradation to indole glucosinolates (38%) as compared to aliphatic glucosinolates (8%) and canning are the most severe heat treatment that may result in significant thermal degradation (73%) of the total amount of glucosinolates [13]. The increasing levels of soluble solids in the recipes resulted in non significantly levels of acidity, anthocyanin, lightness (L), blueness

(a) increases of redness (a) and decreasing firmness (Table 10). The data in table 10 was obtained by increases using processing water from the previous experiments considered as constant using 7000 g cabbage and 2000 g brine composed of a mixture of normal brine as described above and brine with cooking water from previous cooking of red cabbage. Four hundred g cabbage with soluble solids 9.0 g 100 g⁻¹, pH 4.0 were applied and combined with cooking water containing 0 to 900 g cooking water and from 900 to g gram tap water. Five samples of 5000 g cut cabbage 6.8 mm were cooked in 20 l water for 20, 30, 40, 50 and 60 min 260 g cooked cabbage were packed with 345 g brine kept at 70°C and pasteurised at 85°C for and stored at 12 °C for 3 months until analysis After removal of the outer leaves by hand were the remaining leaves cut by hand into 0.5 cm pieces and 2000 g cabbage cuttings were cooked in a stainless steel water bath with 7000 g tap water. After cooking were the cabbage pieces cooled in tap water cooled to 75 °C and transferred into 580 cm³ glass jars cold water net and cooled to 75°C in a nylon net transferred from the cooking pot cooled using in tap water pieces cooled pooling cooking pot. The raw materials from this experiment included application of cabbage leaves from the cultivar 'Aurore'. Five kg cabbage heads from each cultivar without stalk were cut randomly into 6.8 mm slices, mixed carefully and 5000 g cabbage cuttings were heated in 20 l tap water kept at 94 °C for 20, 30, 40 50 and 60 min. After cooling to 12°C in tap water were three samples with 260 g cooked cabbage slices packed in 580 mm glass jars with 345 g brine kept at 70° C. And finally were the glass jars cooled for 5 min in tap water at 60°C and 30 min in tap water until storage at 12°C for three months. Samples of raw cabbage cuttings stored at -25°C until red cabbage study shows all anthocyanins are not created equal analysis. The brine were processed from 2275 g sugar, 65 g acetic acid, K sorbate 6.5 g 20 w/w% acetic acid, 13.5 g w/w Na-benzoate and 4140 g tap water. The jars were rotated by hand once each week during the first month of storage. Data from storage of red cabbage pieces from the cultivars 'Marne' and 'Ruby Perfection' from one harvest time 13th September and 'Aurore' six times from 27 September to 13 December. A number of epidemiological studies have identified an inverse association between consumption of vegetables and the risk of colon and rectal cancer. Animal studies have shown that changes in enzyme activities cause damage resulting from consumption of *Brassica* vegetables or isothiocyanates, that are the breakdown products in the human body. Mechanistic studies have begun to identify the ways in which the compounds may exert their protective action but the relevance of these studies to protective effects in the human alimentary tract is as yet unproven. Studies with a number of specific isothiocyanates have suggested mechanisms that might be the basis of their chemoprotective effects [4]. The concentration and composition of the GLSs in different plants, but also within a plant in the seeds, roots or leaves, may vary greatly and cause changes during plant development. Total aliphatic and indole glucosinolates were significantly modified by all cooking treatments but not by steaming. In general, the steaming led to the lowest loss of total glucosinolates, while stir-frying and stir-frying/boiling presented the highest loss. Stir-frying and stir-frying-boiling are the two most popular methods for most homemade dishes that may cause great losses of chlorophyll, soluble protein, soluble sugar, vitamin C, and glucosinolates, but the steaming method appears to be the best in retention of the nutrients in cooking broccoli. During the food production chain from sowing, picking [16], storage and processing may the composition of glucosinolates change significantly [22], 23]. The results showed that all cooking treatments, except steaming, caused significant losses of chlorophyll and vitamin C and significant decreases of total soluble proteins and soluble sugars. Total aliphatic and indole glucosinolates were significantly modified by all cooking treatments but not by steaming. In general, the steaming led to the lowest loss of total glucosinolates, while stir-frying and stir-frying or boiling presented the highest losses. Stir-frying and stir-frying/boiling, the two most popular methods for most homemade dishes in China may cause great losses of chlorophyll, soluble protein, soluble sugar, vitamin C, and glucosinolates, but the steaming method appears to be the best in retention of the nutrients in cooking broccoli. The various methods used by processing of red cabbage may result in severe significantly reductions of bioactive compounds including carotenoids, anthocyanins and phenolic compounds [15]. Cooking techniques improve the g and freezing of selected cruciferous levels of bioactive compounds and antioxidant activity in kale and red cabbage. Blanching and frozen storage of cauliflower resulted in significantly decreases in aliphatic and indole glucosinolates by 31 and 37 percent and the contents of l-ascorbic acid 19 percent, total phenols 15, anthocyanins were reduced by 19, 15, 38, 16, and 28 per cent [9]. Effects of blanching, boiling of brussel sprouts, white and green cauliflower, broccoli, and curly kale on their contents of glucosinolates contents were determined. It was found that blanch-

ing and cooking of these vegetables led to considerable losses of total glucosinolates from 2.7 to 30.0 per cent and from 35.3 to 72.4 per cent, respectively [9].

Table13. Effects of acetic acid on soluble solids, anthocyanin, firmness, cabbage and colour.

Acetic acid w/w%	Soluble solids g 100 g ⁻¹	Anthocyanin. mg 100 g ⁻¹	pH	Firmness kg	Cabbage g 100 g ⁻¹	L	Blueness	Redness
0.3a	20.6a	63.6a	4.4a	97.6a	81.5a	15.6a	6.2f	-0.8f
0.5a	20.7a	46.3a	4.1b	91.5b	80.9a	16.86b	9.1e	-0.15e
0.7a	20.6a	47.6a	4.0c	89.3d	80.5b	16.4c	10.9d	0.7d
0.9a	20.6a	46.3a	3.8d	90.8c	78.6c	16.9d	10.7c	0.5c
1.1a	20.8a	46.3a	3.8d	87.2d	79.1c	17.3e	12.7d	1.1b
1.3a	21.3a	46.3a	3.8d	76.0e	79.1c	17.2f	13.5a	1.4a

The data in table 14 shows that increasing cooking time not affected pH, acetic acid, anthocyanin, cabbage or lightness, whereas firmness and blueness decreased and redness increased significantly and table 8 shows the results from linear regression analyses. On the basis of this it was concluded that the most important changes by processing are that firmness decreases with increasing acetic acid, increasing soluble solids and processing time. Greenness and yellowness increased by increasing acetic acid. Anthocyanin and yellowness decreased and increased respectively by processing (Table 14). The concentration and composition of the glucosinolates in different plant tissues, such as seeds, roots or leaves may occur during plant development [24]. Furthermore, the effects of various factors in the supply chain of Brassica vegetables including breeding, cultivation, storage and processing on intake and bioavailability of GLSs must be discussed extensively. On the basis of these animal studies have it been shown that changes in enzyme activities may cause damage resulting from consumption of isothiocyanates, that are the breakdown products in the human body. These results in identification of ways in which the relevance of these studies to protective effects in the human alimentary tract has to be considered on the basis of more research [24]. Increases in the sugar content, constant pH, acetic acid, anthocyanin, cabbage, lightness resulted in decreasing firmness [5].

Table14. Cabbage quality characteristics of converted and non-converted samples.

	Contents	Non converted					Converted				
		a	b	c	d	e	a	B	c	d	e
Weight, g		850	850	850	850	850	850	850	850	850	850
Brine, %		15.2e	16.9d	18.7c	25.5b	24.8a	16.8b	22.1a	20.1a	16.2b	14.4c
Firmness, kg	Cabbage	0.9a	71d	74c	84b	89a	126a	113b	108c	103d	100e
Acidity g 100 g ⁻¹	Cabbage	0.4a	0.5a	0.6a	0.4a	0.5a	0.3a	0.4a	0.4a	0.4a	0.4a
“	Brine	0.5a	0.5a	0.4a	0.4a	0.5a	0.3a	0.3a	0.3a	0.4a	0.4a
Sol.sol., g 100 g ⁻¹	Cabbage	16.6e	17.7d	19.2c	20.2b	22.2a	17.6b	17.3b	17.9b	18.4a	18.5a
“	Brine	17.1e	18.0d	19.5c	21.3b	22.6a	17.7c	18.2b	18.6b	18.9b	19.4a
Antho. mg 100g ⁻¹	Cabbage	40.2a	38.1b	35.7c	32.2d	28.5e	39.4a	39.5a	38.7a	39.2a	39.4a
“	Brine	34.9a	33.7a	31.5a	28.7b	25.6b	34.7a	35.3a	34.9a	32.7a	36.2a
Greenness, a	Cabbage	16.1c	18.2b	18.4b	20.3a	20.7a	17.3a	18.6a	17.3a	17.0a	16.9a
Yellowness, b	Cabbage	-3.3e	-2.4d	-2.2c	-1.1b	-1.7a	-4.0b	-3.6a	-3.3a	-3.4a	-3.3a

Effects of cooking time on firmness and surface colour. The results from this research showed that firmness decreased significantly with increasing acetic acid concentration whereas both redness and yellowness increased with increasing concentration of acetic acid. Firmness) = 100.8 - 16.5 acetic acid; r = 0.81; redness a = 5.09 + 6.9acetic acid; r = 0.96 and yellowness b = -1.33 + 2.2acetic acid; r = 0.99. Increasing cooking time resulted in constant sugar, pH, and acetic acid, cabbage and lightness, while anthocyanin, firmness and redness (α) decreased significantly and yellowness increased significantly during cooking. During the food production chain from sowing, picking, storage and processing may the composition of glucosinolates change significantly [4,16, 22, 23]. Decreasing greenness and yellowness resulted in significantly decreasing firmness, decreases in redness and increasing yellowness (Table 15). Table 14 shows the results from using previously liquids by processing of red cabbage pieces with solids 9.0 g at pH 3.97. Four hundred g red cabbage pieces with pH 3.97 and processing liquids from previous cooking of red cabbage. 400 g cabbage was boiled in 900 g boiling water containing from 0 to 900 g processing liquids and 900 to 0 g tap water. The contents of anthocyanin were 38.8 mg 100 g⁻¹ in the cooking liquids. Table 14 shows that the cooking water content of

anthocyanin in the cooking water increased from 0 to 38.8 mg 100 g⁻¹anthocyanin and the first cooked red cabbages contained 3.4 g 100 g⁻¹. Soluble solids and the contents of 24.0 mg 100 g⁻¹ and 3.4 g 100 g⁻¹ soluble solids. After ten cookings with ten supplements contained the last cooking of 7.8 g 100 g⁻¹ soluble solids and 40.5 mg 100 g⁻¹ anthocyanin. Lightness, greenness decreased while yellowness increased due to the contents in the supplied samples. ΔE and firmness increases, pH was constant and the contents of soluble solids and anthocyanin increased significantly with increasing sample number.

Table15. Effects of increasing cooking time on quality characteristics.

Cooking min	Sugar g 100 g ⁻¹	pH	Acetic acid g 100 g ⁻¹	Anthocyanin mg 100 g ⁻¹	Firmness kg	Cabbage g 100 g ⁻¹	L	a	b
10	20.8a	4.0a	0.6a	39.5a	361.0a	64.6a	17.0a	14.1a	-0.4a
15	20.9a	4.0a	0.6a	40.1b	256.0b	66.4a	17.0a	14.4a	-0.1a
25	20.9a	4.0a	0.6a	39.2a	223.5c	67.1a	17.1a	14.5a	-0.1b
30	21.3b	4.0a	0.6a	37.4c	112.5d	68.3a	16.3b	11.9b	0.2b
40	21.2b	4.0a	0.6a	34.6c	89.8c	67.3a	17.2a	13.4b	0.9b
50	21.4b	4.0a	0.6a	35.8c	75.4f	68.0a	17.4a	12.0b	0.6b
60	20.1a	4.0a	0.6a	30.9d	61.6g	68.7a	18.3a	11.4b	0.4a

The data in table 16 shows that the composition of two samples of red cabbage and brine were completely equilibrated. A number of epidemiological studies have identified an inverse association between consumption of these vegetables and the risk of colon and rectal cancer. Animal studies have shown that changes in enzyme activities may cause damage resulting from consumption of *Brassica* vegetables or isothiocyanates that are the breakdown products in the human body. Mechanistic studies have begun to identify the ways in which the compounds may exert their protective action but the relevance of these studies to protective effects in the human alimentary tract is as yet unproven. Studies with a number of specific isothiocyanates have suggested mechanisms that might be the basis of their chemo protective effects. The concentration and composition of the GLSs in different plants, but also within a plant in the seeds, roots or leaves, can vary greatly and also changes during plant development. Furthermore, the effects of various factors in the supply chain of *Brassica* vegetables including breeding, cultivation, storage and processing on intake and bioavailability of GLSs are discussed in this paper. The contents of ingredients were completely equalized regarding acidity 0.72 g 100 g⁻¹, soluble solids 24.8 g 100 g⁻¹ and anthocyanin 27.5 mg 100 g⁻¹. The average drained weight (Dw) decreased significantly linearly with cooking time Dw = 313-0.123min. r = 0.98 and the average firmness decreased exponentially with cooking time. Total aliphatic and indole glucosinolates were significantly modified by all cooking treatments but not by steaming. In general, the steaming led to the lowest loss of total glucosinolates, while stir-frying and stir-frying/boiling presented the highest loss. Stir-frying and stir-frying/boiling are the two most popular methods for most homemade dishes that may cause great losses of chlorophyll, soluble protein, soluble sugar, vitamin C, and glucosinolates, but the steaming method appears to be the best in retention of the nutrients in cooking broccoli. Processing of red cabbage by increasing contents of acetic acid at constant levels of soluble solids, anthocyanin and pH resulted in decreasing firmness and cabbage, whereas lightness, blueness, and redness increased.

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