

Synthesis, Structure, and Biological Screening of Some Condensed Heterocyclic Compounds from 8-Chlorotetrazolo [5,1-*f*]-1,2,4-Triazine Precursor

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Abstract: 8-Chlorotetrazolo [5,1-*f*]-1,2,4-triazine (**2**) reacted with different compounds to yield a variety of novel condensed heterocyclic nitrogen compounds. The newly prepared structures were characterized by spectral data and screened for their antimicrobial activities against bacteria and fungi strains.

Keywords: 8-Chlorotetrazolo [5,1-*f*]-1,2,4-triazine, reactions, condensed heterocyclic compounds, antimicrobial assessment

1. INTRODUCTION

Many compounds consisting of the tetrazole nucleus have received¹⁻¹⁰ much great attention because of their wide range of therapeutic and biological properties.^{11,12} They have emerged as antibacterial,^{2-8,13} antiproliferation,¹⁴ anticancer,¹⁴ and anticonvulsant¹⁵ agents. In this article, it is our intention to enlarge the area of the investigation towards tetrazolo-heterocycles using framework different reagents and expected interesting antimicrobial agents.

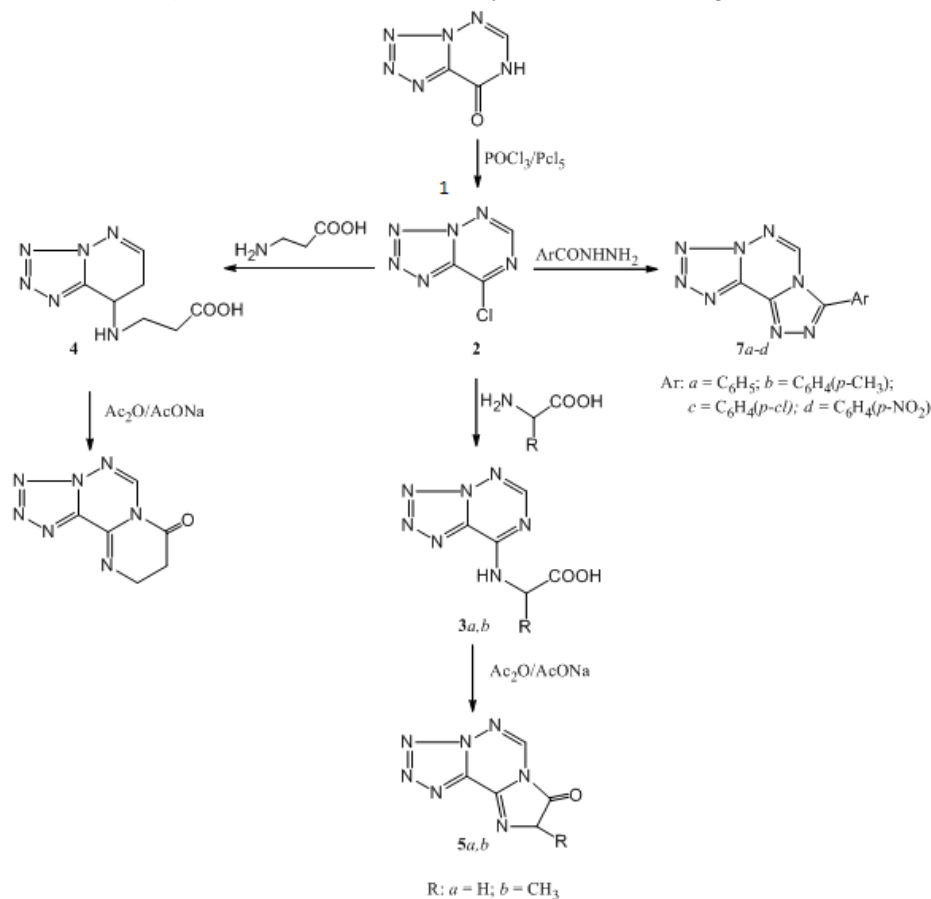
2. RESULTS AND DISCUSSION

The IR spectrum of tetrazolo[5,1-*f*]-1,2,4-triazin-8(7H)-one⁴(**1**) showed a characteristic absorption bands at 3260 and 1670 cm⁻¹ corresponding to the NH and CONH groups and the ¹H NMR spectrum revealed exchangeable NH signal at 11.81 ppm. Compound **1** was treated with a mixture of phosphorous pentachloride and phosphorous oxychloride to yield the corresponding 8-chlorotetrazolo[5,1-*f*]-1,2,4-triazine (**2**) and was confirmed on the basis of its elemental analysis and devoid any bands for NH and CONH groups in IR region. The structure of compound **2** is a promising for the synthesis of diverse condensed heterocyclic nitrogen compounds (Schemes **1** and **2**).

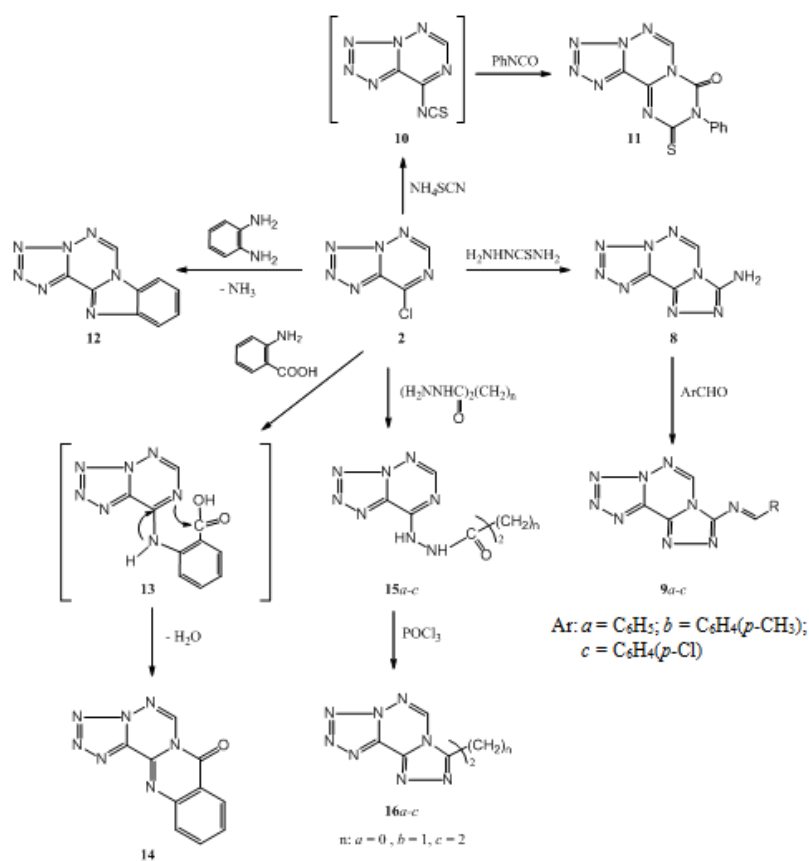
Treatment of cyclic imidoyl chloride **2** with the sodium salt of various amino acids namely: glycine, D-alanine, and β-alanine under reflux conditions produced the corresponding tetrazolotriazinylamino acids **3a,b** and **4**. The IR spectra of the latter compounds were confirmed on the basis absorption for OH, NH and CO groups. The ¹H NMR spectrum of compound **3a** displayed CH₂ protons as a singlet signal at 4.56 ppm and exchangeable OH and NH protons as singlet signals at 11.21 and 8.12 ppm, respectively. The ¹H NMR spectrum of compound **3b** displayed a CH proton as a quartet signal at 4.42 ppm and CH₃ protons as a doublet signal at 1.51 ppm. The ¹H NMR spectrum of compound **4** showed a triplet signal at 3.62 ppm assigned for CH₂CO and triplet signal at 3.22 ppm assigned for NCH₂ beside exchangeable OH and NH protons. The mass spectra of **3a,b** and **4** revealed the correct molecular ions which were supported by elemental analysis. The amino acid derivatives **3a,b** and **4** were easily cyclized¹⁶ via 1,3 tautomerism in heating acetic anhydride in the presence of anhydrous sodium acetate to give imidazotetrazolotriazine derivatives **5a,b** and tetrazolopyrimidinotriazine **6**. The IR region of these compounds displayed the disappearance OH, NH and CO absorptions and showed of the absorption bands for amide groups at 1670-1680 cm⁻¹. The ¹H NMR spectra of compounds **5a,b** and **6** showed the absence of OH and NH signals characteristic of the parent amino acids **3a,b** and **4**. These data together with the correct elemental analysis are compatible with **5a,b** and **6** structures.

The reaction of chloro compound **2** with variety of aromatic acid hydrazides namely: benzoic, *p*-toluic, *p*-chlorobenzoic, and *p*-nitrobenzoic afforded 3-aryl-1,2,4-triazolo[4,3-*d*]tetrazolo [5,1-*f*]-1,2,4-triazine derivatives **7a-d**, respectively. Elemental and spectra data of the latter compounds are consistent with the structure assigned to their compounds (*cf.* Experimental). More recently in the

literature,⁴ the aforementioned triazolotriazinone structures were constructed by the reaction of 8-hydrazinotetrazole [5,1-*f*]-1,2,4-triazine with triethyl orthoformate or glacial acetic acid.



Scheme 1



Scheme 2

Cyclization of cyclic imidoyl chloride **2** using thiosemicarbazide in ethanol yielded 3-amino-1,2,4-triazolo derivative **8**. The ^1H NMR spectrum of compound **8** displayed two protons assigned exchangeable NH_2 as a singlet signal at 6.30 ppm. Additionally, the condensation of **8** with different aromatic aldehydes namely: benzaldehyde, *p*-tolualdehyde, and *p*-chlorobenzaldehyde leading to the arylidenes **9a-c**, respectively. The IR spectra of the latter compounds possessed a characteristic absorption bands at 1620 and 1628 cm^{-1} corresponding to the $\text{C}=\text{N}$ groups. The ^1H NMR spectra of compounds **9a-c** revealed the presence azomethine ($\text{CH}=\text{N}$) at δ 8.52 and 9.02 ppm.

Furthermore, the reaction of chloro compound **2** with ammonium thiocyanate in ethanol produced the unisolable intermediate **10** that reacted *in situ* with phenyl isocyanate *via* 2+4 cycloaddition reaction to build 3-phenyl-2-thioxo-2,3-dihydro-tetrazolo[5',1':6,1]-1,2,4-triazino[4,5-*a*]-1,3,5-triazin-4-one(**11**). The IR spectrum of **11** showed the presence absorption bands at 1665 and 1270 cm^{-1} attributed to CON and $\text{C}=\text{S}$, respectively.

The compound **2** and *o*-phenylenediamine dihydrochloride were successfully cyclized through the elimination of an ammonia molecule to the corresponding tetrazolo[5',1':6,1]-1,2,4-triazino[4,5-*a*]benzo[*d*]imidazole (**12**). Both IR and ^1H NMR spectra showed no signals corresponding to the NH and NH_2 groups thus, confirming the structure of compound **12** whereas its mass spectrum showed the molecular ion peak at $m/z = 210.86$ which was in agreement with molecular formula $\text{C}_9\text{H}_5\text{N}_7$ ($m/z=211$).

Also, interaction of compound **2** with anthranilic acid under fusion conditions, the expected quinazolinone derivative **14** was resulted as the only isolable product. The formation of **14** was explained by the formation of unisolable intermediate **13**, undergoes intramolecular ring closure to form the exactly product **14**, which IR and ^1H NMR spectra exhibited devoid and OH and NH groups but showed CON absorption at 1670 cm^{-1} in the IR region. The mass spectrum of **14** showed a peak corresponding to its molecular ion at $m/z=239$ ($\text{C}_{10}\text{H}_5\text{N}_7\text{O}$).

Moreover, treatment of two molar equivalents of cyclic imidoyl chloride **2** with one molar equivalent of each oxalic, malonic, and succinic acid dihydrazides caused a product in each case whose structure was verified from spectroscopic data. Its IR spectrum showed absorptions characteristic of 2NH and CON as well as two singlets attributed to 2NH each (exchangeable with D_2O) ^1H NMR signals. Accordingly, the previously products were decisively assigned as bishydrazides **15a-c**. The aforementioned products were in accordance with previous report⁵ with cyclic imidoyl chlorides.

On the other hand the structures **15a-c** underwent cyclodehydration by phosphorus oxychloride to construct the bistriazolotetrazolotriazine derivatives **16a-c** which showed IR absorptions characteristic of only a $\text{C}=\text{N}$ and, most importantly, lacked any NH and CON absorption bands characteristic of the parent compounds **15a-c**. Furthermore, the mass spectra of **16a-c** showed molecular ions in agreement with the assigned structures. Reaction¹⁷ of cyclic imidoyl chloride **2** with acid dihydrazides directly gave bistriazolo-structures.

3. ANTIMICROBIAL SCREENING

The newly prepared compounds were screened *in vitro* for their antimicrobial properties against Gram-positive (*Staphylococcus aureus* and *Bacillus subtilis*) and Gram-negative (*Klebsiella pneumoniae* and *Escherichia coli*) bacteria strains and (*Aspergillums niger* and *Candida albican*) fungi strains. Ampicillin and Clotrimazole were used as standard drugs for bacteria and fungi, respectively. The minimal inhibitory concentration¹⁸ (MIC , in $\mu\text{g. cm}^{-3}$), and the results are summarized in **Table 1** showing that **4**, **6**, **7a**, **9a** and **12** exhibit an antimicrobial activity against *S. aureus* (25%); **3b**, **6**, **11**, **12** and **16a** against *B. subtilis* (25%); **7a**, **8**, **14**, **15a** and **16c** against *K. pneumoniae* (50%); **5**, **6**, **11**, **12** and **16a** (50%) while compound **3b** possessed activity against *E. coli* comparable to that ampicillin. Moreover, **3a**, **7c**, **8** and **14** possessed an antimycotic activity against *A. niger* (50%), and **4**, **5**, **11** and **12** against *C. albican* (50%) comparable to that clotrimazole. The compound of **15c** showed lower activity than the reference standards (ampicillin and clotrimazole) against the test organisms.

Table1. Antimicrobial activity of synthesized compounds ($MIC/\mu g\ cm^{-3}$)

Compound	Bacterial strain				Fungal strain	
	<i>S.aureus</i>	<i>B.subtitis</i>	<i>K.pneumoniae</i>	<i>E.coli</i>	<i>A.niger</i>	<i>C.albican</i>
3a	100	>200	100	>200	25	100
3b	100	50	100	25	100	100
4	50	100	100	100	100	25
5	>200	100	100	50	100	25
6	50	50	100	50	100	100
7a	50	100	50	>200	100	100
7c	100	>200	>200	100	25	100
8	100	100	50	100	25	100
9a	50	100	100	100	100	100
11	100	50	100	50	100	25
12	50	50	100	50	100	25
14	100	100	50	100	25	100
15a	100	100	50	100	100	100
15c	100	100	100	100	100	100
16a	100	50	100	50	100	100
16c	100	100	50	>200	100	100
Ampicillin	12.5	12.5	25	25	--	-
Clotrimazole	--	--	--	--	12.5	12.5

4. EXPERIMENTAL

4.1. General

Melting points were measured with a Gallenkamp apparatus and are uncorrected. The reactions were followed up and the purification of products was carried out on pre-(layer thickness 0.25mm; coated TLC plates Silica Gel-Merck), visualizing the spots in Iodine. IR spectra were recorded (KBr) on a Shimadzu FT-IR 8101 PC infrared spectrophotometer. The 1H NMR spectra were determined in DMSO(d_6) at 300 MHz on a Varian Mercury VX 300 NMR spectrometer and their chemical shifts (δ /ppm) are reported using TMS as internal standard. Mass spectra were recorded on a HP model MS 5988 spectrometer at electron ionizing energy of 70 eV. Elemental analyses were performed by the microanalytical Unit, Cairo University, Egypt.

4.2. 8-Chlorotetrazolo[5,1-f]-1,2,4-triazine (2)

A suspension of tetrazolo[5,1-f]-1,2,4-triazin-8(7H)-one⁴ (**1**, 0.006 mol) and phosphorous pentachloride (0.006 mol) in phosphorous oxochloride (10 cm³) was heated under reflux on a water-bath for 2h. Then the mixture was cooled to room temperature and poured into crushed ice-water slowly. The obtained solid was filtered off, washed with cold water, dried and recrystallized from abs. ethanol to give **2**, yield 0.84g (74.34%); m.p. 190-192 °C; 1H NMR (DMSO- d_6) δ : 6.25 ppm (s, 1H, CH); MS: m/z (%) = 156 ($M^+ Cl^{35}$, 11)

Anal. Calcd. For C₃HClN₆(156.5): C, 23.00; H, 0.64; N, 53.67. Found: C, 22.61; H, 1.04; N, 53.34%.

4.3. General procedure for the synthesis of tetrazolotriazinyl amino acids 3a,b and 4

The corresponding amino acids (0.006 mol) namely: glycine, D-alanine or β -alanine and sodium carbonate (0.06 mol) were dissolved in water (20 cm³). Then compound **2** (0.006 mol) was added to it and refluxed for 6h. The reaction mixture was left overnight at ambient temperature, and then treated with cold hydrochloric acid. The separated product was collected by filtration, dried and crystallized from abs. ethanol. The physico-chemical and spectra data of **3a,b** and **4** the following:

4.4. 2-(Tetrazolo[5,1-f]-1,2,4-triazin-8-ylamino)acetic acid (3a)

Yield: 0.96g (77.05%); m.p. 270-272 °C; IR (KBr): ν =3300 (OH), 2445 (NH), 1720 cm⁻¹ (CO); 1H NMR (DMSO- d_6): δ = 4.56 (s, 2H, CH₂), 6.26 (s, 1H, CH), 8.12 (s, 1H, NH, exchangeable with D₂O), 11.21 ppm (s, 1H, OH exchangeable with D₂O); MS: m/z (%), 196 ($M^+ + 1$, 16).

Anal. Calcd. for C₅H₅N₇O₂ (195): C, 30.77; H, 2.56; N, 50.26. Found: C, 31.11; H, 2.41; N, 50.11%.

4.5. 2-(Tetrazolo[5,1-f]-1,2,4-triazin-8-ylamino)propanoic acid (3b)

Yield: 0.93g (69.66%); m.p. 260-261 °C; IR (KBr): ν =3400 (OH), 2430 (NH), 1705 cm⁻¹ (CO); 1H NMR (DMSO- d_6): δ = 1.51 (d, 3H, CH₃), 4.42 (q, 1H, CH), 6.20 (s, 1H, CH), 8.22 (s, 1H, NH,

exchangeable with D₂O), 10.83 ppm (s, 1H, OH exchangeable with D₂O); MS: m/z (%) = 211 (M⁺+2, 20).

Anal. Calcd. For C₆H₇N₇O₂ (209): C, 34.45; H, 3.35; N, 46.89. Found: C, 34.49; H, 3.31; N, 47.01%.

4.6.3-(Tetrazolo[5,1-f]-1,2,4-triazin-8-ylamino)propanoic acid (4)

Yield: 0.88g (65.92%); m.p. 268-270 °C; IR (KBr): ν =3300 (OH), 2395 (NH), 1720 cm⁻¹ (CO); ¹H NMR (DMSO-*d*₆): δ = 3.22 (t, 2H, NCH₂), 3.62 (t, 2H, CH₂CO), 6.25 (s, 1H, CH), 8.24 (s, 1H, NH, exchangeable with D₂O), 11.02 ppm (s, 1H, OH exchangeable with D₂O); MS: m/z (%) = 209 (M⁺, 6).

Anal. Calcd. For C₆H₇N₇O₂ (209): C, 34.45; H, 3.35; N, 46.89. Found: C, 34.55; H, 3.48; N, 46.85%.

4.7. General procedure for the synthesis of imidazotetrazolotriazines 5a,b and tetrazolo pyrimidotriazine 6

A mixture of the appropriate (3*a,b* or 4, 0.005 mol) and acetic anhydride (20 cm³), and anhydrous sodium acetate (0.42g, 0.005 mol) was heated under reflux for 4h. The solvent was removed under reduced pressure, the obtained residue was washed with water the recrystallized from abs. ethanol. The following data of the title compounds were prepared according to this procedure are described:

4.8. Imidazo[1,2-d]tetrazolo[5,1-f]1,2,4-triazin-3(2H)-one (5a)

Yield: 0.58g (63.74%); m.p. 250-251 °C; IR (KBr): ν =1670 cm⁻¹ (CON); ¹H NMR (DMSO-*d*₆): δ = 4.66 (s, 2H, CH₂), 6.20 ppm (s, 1H, CH), MS: m/z (%) = 179 (M⁺+2, 8).

Anal. Calcd. for C₅H₃N₇O (177): C, 33.90; H, 1.70; N, 55.37. Found: C, 33.81; H, 1.74; N, 55.41%.

4.9. 2-Methylimidazo[1,2-d]tetrazolo[5,1-f]-1,2,4-triazin-3(2H)-one (5b)

Yield: 0.62g (68.13%); m.p. 240-242 °C; IR (KBr): ν =1675 cm⁻¹ (CON); ¹H NMR (DMSO-*d*₆): δ = 1.62 (d, 3H, CH₃), 4.60 (q, 1H, CH), 6.22(s, 1H, CH) MS: m/z (%) = 191 (M⁺, 6).

Anal. Calcd. for C₆H₅N₇O (191): C, 37.70; H, 2.62; N, 51.31. Found: C, 38.01; H, 2.44; N, 51.52%.

4.10. Tetrazolo[5,1-f]pyrimido[1,2-d]-1,2,4-triazin-4(2H, 3H)-one (6)

Yield: 0.52g (57.14%); m.p. 245-246 °C; IR (KBr): ν =1680 cm⁻¹ (CON); ¹H NMR (DMSO-*d*₆): δ = 3.10 (t, 2H, NCH₂), 4.05 (t, 2H, COCH₂), 6.28 ppm (s, 1H, CH) MS: m/z (%) = 191 (M⁺, 13).

Anal. Calcd. for C₆H₅N₇O (191): C, 37.70; H, 2.62; N, 51.31. Found: C, 37.42; H, 3.01; N, 51.60%.

4.11. Reaction of 8-chlorotetrazolo[5,1-f]-1,2,4-triazine (2) with some acid hydrazides

4.11.1. General Procedure

A mixture of compound (2, 0.006 mol) and appropriate acid hydrazides (0.006 mol) namely: benzoic, *p*-toluic; *p*-chlorobenzoic, and *p*-nitrobenzoic in ethanol (30 cm³) was refluxed for 3h. after cooling the mass product was filtered off and recrystallized from abs. ethanol to provide 7*a-d*.

4.12. 3-Phenyl-1,2,4-triazolo[4,3-d]tetrazolo[5,1-f]-1,2,4-triazine (7a)

Yield: 0.91g (59.83%); m.p. 210-211 °C; IR (KBr): ν =1620 cm⁻¹ (C=N); ¹H NMR (DMSO-*d*₆): δ = 6.20 (s, 1H, CH), 7.02-8.52 ppm (m, 5H, ArH), MS: m/z (%) = 240 (M⁺+2, 21).

Anal. Calcd. for C₁₀H₆N₈ (238): C, 50.42; H, 2.52; N, 47.06. Found: C, 50.55; H, 2.31; N, 46.82%.

4.13. 3-*p*-Tolyl-1,2,4-triazolo[4,3-d]tetrazolo[5,1-f]-1,2,4-triazine (7b)

Yield: 0.84g (52.17%); m.p. 225-226 °C; IR (KBr): ν =1625 cm⁻¹ (C=N); ¹H NMR (DMSO-*d*₆): δ = 2.30 (s, 3H, CH₃), 6.25(s, 1H, CH), 7.10-8.61 ppm (m, 4H, ArH), MS: m/z (%) = 252 (M⁺, 17).

Anal. Calcd. for C₁₁H₈N₈ (252): C, 52.38; H, 3.18; N, 44.44. Found: C, 52.11; H, 3.22; N, 44.49%.

4.14. 3-*p*-Chlorophenyl-1,2,4-triazolo[4,3-d]tetrazolo[5,1-f]-1,2,4-triazine (7c)

Yield: 0.94g (54.02%); m.p. 240-241 °C; IR (KBr): ν =1630 cm⁻¹ (C=N); ¹H NMR (DMSO-*d*₆): δ = 6.20 (s, 1H, CH), 7.17-8.51 ppm (m, 4H, ArH), MS: m/z (%) = 237 (M⁺ Cl³⁵, 8).

from abs. ethanol to produce **12**. yield: 0.82g (65.08%); m.p. 210-212 °C; ¹H NMR (DMSO-*d*₆): δ = 6.24 (s, 1H, CH), 7.20-8.12 ppm (m, 4H, ArH); MS: *m/z* (%) = 210.86 (M⁺, 26).

Anal. Calcd. for C₉H₅N₇ (211): C, 51.18; H, 2.37; N, 46.45. Found: C, 50.98; H, 2.11; N, 46.31%.

4.23. Tetrazolo[5⁻,1⁻:6,1]-1,2,4-triazino[4,5-*b*]quinazolin-8-one (14)

A mixture of **2** (0.006 mol) and anthranilic acid (0.006 mol) was heated in oil bath at 190-191 °C for an hour and then allowed to cool down at room temperature; the obtained residue was crystallized from abs. ethanol to result **14**. yield: 0.94g (61.44%); m.p. 220-222 °C; IR (KBr): ν=1670 (CON), ¹H NMR (DMSO-*d*₆): δ = 6.27 (s, 1H, CH), 7.05-8.10 ppm (m, 4H, ArH); MS: *m/z* (%) = 239 (M⁺, 22).

Anal. Calcd. for C₁₀H₅N₇O (239): C, 50.21; H, 2.09; N, 41.00. Found: C, 49.91; H, 2.11; N, 41.33%.

4.24. Synthesis of 15a-c: General Procedure

To a solution of **2** (0.006 mol) in 15 cm³ abs. ethanol, a solution of (0.006 mol) corresponding acid dihydrazide namely: oxalic, malonic and succinic in 15 cm³ abs, ethanol was gradually added, and the mixture was heated under reflux at 100 °C for an hour. The product which separated upon cooling was filtered off, and crystallized from mixture of water and abs. ethanol to afford the products **15a-c**.

4.25. Oxalyl bis{(tetrazolo[5,1-f]-1,2,4-triazin-8-yl) hydrazide} (15a)

Yield: 1.67g (72.93%); m.p. 230-232 °C; IR (KBr): ν=3285, 3280 (2NH), 1665 cm⁻¹ (CON), ¹H NMR (DMSO-*d*₆): δ = 6.15, 6.20, (2s, 1H each, 2CH); 11.90, 12.20 ppm (2s, 2H each, 2 NH each, exchangeable with D₂O); MS: *m/z* (%) = 360 (M⁺+2, 27).

Anal. Calcd. for C₈H₆N₁₆O₂ (358): C, 26.82; H, 1.68; N, 62.57. Found: C, 27.25; H, 1.71; N, 62.67%.

4.26. Malonyl bis{(tetrazolo[5,1-f]-1,2,4-triazin-8-yl) hydrazide} (15b)

Yield: 1.45g (60.92%); m.p. 220-222 °C; ¹H NMR (DMSO-*d*₆): δ = 4.41 (s, 2H, CH₂); 6.20 (s, 2H, 2CH), 11.85, 12.30 ppm (2s, 2H each, 2NH each, exchangeable with D₂O); MS: *m/z* (%) = 372 (M⁺, 10).

Anal. Calcd. for C₉H₈N₁₆O₂ (372): C, 29.03; H, 2.15; N, 60.22. Found: C, 28.86; H, 2.22; N, 60.52%.

4.27. Succinyl bis{(tetrazolo[5,1-f]-1,2,4-triazin-8-yl) hydrazide} (15c)

Yield: 1.34g (54.25%); m.p. 240-243 °C; IR (KBr): ν=3275, 3290 (2NH), 1690 cm⁻¹ (CON), ¹H NMR (DMSO-*d*₆): δ = 3.25, 3.40 (2t, 2H each, CH₂ each); 6.25 (s, 2H, 2CH), 12.01, 12.20 ppm (2s, 2H each, 2NH each, exchangeable with D₂O); MS: *m/z* (%) = 387 (M⁺ + 1, 9).

Anal. Calcd. for C₁₀H₁₀N₁₆O₂ (386): C, 31.09; H, 2.59; N, 58.03. Found: C, 30.82; H, 2.61; N, 57.82%.

4.28. Synthesis of 16a-c: General Procedure

The respective of **15a-c** (0.006 mol) was treated with 20 cm³ phosphorus oxychloride and heated under reflux for 2h. After attaining ambient temperature, the mixture was poured onto a cold saturated solution of sodium bicarbonate and the crude solid which separated was filtered off, washed with water, dried, and crystallized from abs. ethanol to yield the products **16a-c**.

4.29. Bis {1,2,4-triazolo[4,3-*d*]tetrazolo [5,1-f]-1,2,4-triazin-3-yl} (16a)

Yield: 1.62g (78.64%); m.p. 240-242 °C; IR (KBr): ν=1610 cm⁻¹ (C=N), MS: *m/z* (%) = 323 (M⁺+1, 10).

Anal. Calcd. for C₈H₂N₁₆ (322): C, 29.81; H, 0.62; N, 69.57. Found: C, 29.90; H, 1.05; N, 70.01%.

4.30. Bis {1,2,4-triazolo[4,3-*d*]tetrazolo [5,1-f]-1,2,4-triazin-3-yl} methane (16b)

Yield: 1.72g (80.00%); m.p. 230-232 °C; ¹H NMR (DMSO-*d*₆): δ = 4.46, (s, 2H, CH₂), 6.25 ppm (s, 2H each, 2CH); MS: *m/z* (%) = 336 (M⁺, 27).

Anal. Calcd. for C₉H₄N₁₆ (336): C, 32.14; H, 1.19; N, 66.67. Found: C, 32.09; H, 1.14; N, 67.09%.

4.31. Bis {1,2,4-triazolo[4,3-*d*]tetrazolo [5,1-*f*]-1,2,4-triazin-3-yl} ethane (16c)

Yield: 1.76g (78.57%); m.p. 250-271 °C; IR (KBr): $\nu=1620\text{ cm}^{-1}$ (C=N), MS:

m/z (%) = 351 ($M^+ + 1$, 19).

Anal. Calcd. for $C_{10}H_6N_{16}$ (350): C, 34.29; H, 1.71; N, 64.00. Found: C, 34.09;

H, 1.95; N, 64.24%.

5. ANTIMICROBIAL SCREENING

Solutions of the test compounds, ampicillin trihydrate and clotrimazole were prepared in *DMSO* at a concentration of $100\text{ }\mu\text{g}/\text{cm}^3$. Twofold dilution of the compounds were prepared (800, 400, $6.25\text{ g}/\text{cm}^3$). The microorganism suspensions at 10^6 colony Formin Unit 1 cm^3 (CFU/cm^3) concentration were inoculate to the corresponding wells. Plates were incubated at 36°C for 24 to 28h the incubation chamber. The minimal inhibitory concentration (*MIC*) were determined. Controls with *DMSO* and infected media were also investigated.

6. CONCLUSION

The foregoing results demonstrated the utility of 8-chlorotetrazolo [5,1-*f*]-1,2,4-triazine as synthons for the construction of some condensed heterocyclic nitrogen structures by different cyclization reagents. The antibacterial and antifungi activities of the synthesized compounds were even comparable to ampicillin and clotrimazole.

REFERENCES

- [1] (a) Gein, V. L.; Zamaraeva, T. M.; Dmitriev, M. V., *Russ. J. Org. Chem.*, **2016**, 52 558.
(b) Golobokova, T. V.; Vereshchagin, L. I., Ratovskii, G. V.; Proidakov, A. G.; Kizhnyayev, V. N. *Ibid.* **2016**, 52, 1039.
- [2] El Badry, S. M.; Taha, M. A. M. *J. Korean Chem. Soc.*, **2014**, 58, 381.
- [3] El-Badry, S.M.; Taha, M.A.M. *J. Korean. Chem. Soc.* **2011**, 55, 974.
- [4] Taha, M.A.M.; El-Badry, S.M. *J. Korean. Chem. Soc.* **2010**, 54, 414.
- [5] Taha, M.A.M.; El-Badry, S.M. *Monatsh. Chem.* **2008**, 139, 1261.
- [6] Taha, M.A.M. *Phosphorus, Sulphur, Silicon Relat. Elem.* **2008**, 183, 2525.
- [7] Taha, M.A.M. *Monatsh. Chem.* **2007**, 138, 505.
- [8] Taha, M.A.M.; El-Badry, S.M. *Phosphorus, Sulfur, Silicon Relat Elem* **2007**, 182, 1011.
- [9] Taha, M.A.M.; El-Badry, S.M. *J. Chin. Chem. Soc.* **2006**, 53, 1181.
- [10] Taha, M.A.M. *J. Chin. Chem. Soc.* **2005**, 52, 137.
- [11] Moderhack, D. *J. Prakt. Chem.* **1998**, 340, 687.
- [12] Kolodobskii, G. I.; Ostrovskii, V. A.; Popavskii, V. S. *Chem. Heterocycl. Cpds.* **1981**, 17, 965.
- [13] Karnik, A. V.; Malviya, N.J.; Kulkarni, A.M.; Jadhav, B.L. *Eur. J. Med. Chem.* **2006**, 41, 891.
- [14] Jantova, S.; Ruzekova, I.; Stantovsky, S.; Spirkova, K. *Neoplasma* **1997**, 44, 240.
- [15] Rubat, C.; Coudert, P.; Couquelet, J.M.; Tronche, P.; Bastide, J.; Bastide, P. *Formaco* **1990**, 45, 331.
- [16] Afify, A. A.; El-Nady, S.; Sayed, M. A.; Mohey, I. *Indian J. Chem.* **1988**, 27B, 920.
- [17] Shaban, M. A. E.; Taha, M.A. M.; Nazr, A. Z., *J. Islamic Acad. Sci*, **1990**, 3, 98.
- [18] Murray, P.R.; Baron, E.J.; Pfaller, M.A.; Tenover, F.C.; Tenover, R.T. Manual of Clinical Microbiology. In: *Antimicrobial Agents and Susceptibility Testing*. Woods, G. L., Washington J.A., Eds.; ASM Press: Wahington DC, **1995**, 1327.